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## WHITE PAPERS & APPLICATION NOTES



### Whisky Analysis by Raman Spectroscopy

Raman spectroscopy is an analytical technique which can be used both quantitatively and qualitatively. This application note details the quantitative use of Raman spectroscopy to determine ethanol content in samples of whisky. Qualitatively, Raman spectroscopy can also be used for whisky analysis to ensure it does not contain methanol, a toxic alcohol which can be fraudulently used in alcohol sales to boost profits.

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**APPLICATION NOTE**

**Whisky Analysis by Raman Spectroscopy**  
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**Introduction**

Whisky is a distilled alcoholic beverage named from the Gaelic "uisge beatha" meaning "water of life". The drink is made from fermented grain mashes with different grains used to provide the wide variety of whisky flavours available. When bottles of whisky are sold, they will contain a % alc, this refers to how much ethanol is present in the bottle. Commonly, whisky is 40% alc, thus 40% of the bottle should be ethanol. Whisky is an increasingly popular drink, with the Scotch Whisky Association stating 44 bottles of Scotch Whisky are exported every second, with exports worth £6.7bn in 2021.

"The potential effects from methanol poisoning range from, but are not limited to, central nervous system depression, nausea, blindness, coma, and death."

Determining the % ethanol in bottles of whisky is required to ensure products are being sold authentically and checking there is no methanol present in bottles is critical for keeping products safe for consumption. This application note details how Raman spectroscopy can offer quantitative and qualitative analysis of whisky samples.

**Materials and Methods**

An RBS Raman Microscope equipped with a 785nm excitation laser and a cuvette holder was used for the analysis. Pure ethanol and methanol were purchased from Merck for the standard solutions, whisky samples were sourced locally. To create the calibration curve aqueous solutions of 20%, 40%, 60%, 80%, and 100% of ethanol were made. Raman spectra were taken for each solution and sample using the same measurement conditions: 100% laser power, 65 second exposure time, 300 gr/mm grating, 100 µm slit, 2 mm pinhole.



Figure 1 Scotch Whisky from a glass bottle

Producers sales of drinking alcohol, especially spirits, is a concerning and potentially dangerous system which vendors try to increase their profits. There are two common ways in which alcohol is sold. Legally, firstly is the sale of watered down alcohol. This is either done in bulk, where bottles are sold with fraudulent % alc's printed on them, or in bars where spirits are diluted after wholesale purchase. The second way seen is alcohol fraud is more sinister where methanol is used in spirits, instead of ethanol.

Ethanol metabolises in the liver by a process called alcohol dehydrogenase eventually being broken down to harmless carbon dioxide and water. Methanol meanwhile is toxic to humans, it's commonly found in antifreeze. The two molecules only differ by one carbon and two hydrogen atoms, but after consuming methanol, the alcohol dehydrogenase enzyme converts the methanol into formaldehyde which is then quickly metabolised into formic acid. According to the biological overview from Public Health England, methanol is toxic following ingestion, inhalation, and via dermal exposure.

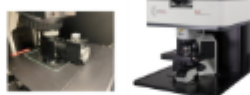


Figure 2 Edinburgh Instruments Raman Microscope

A whisky sample was spiked to create a whisky solution with 10% methanol concentration. A comparison between unspiked whisky, spiked whisky, and methanol was carried out using a 300gr/mm grating to observe the full spectral range in a single snapshot, and with a 1200 gr/mm grating to see higher spectral resolution in the fingerprint region of the spectrum (500 cm<sup>-1</sup> to 1000 cm<sup>-1</sup>).

**Verifying Ethanol Content**

In analytical chemistry a calibration (or standard) curve is a method for determining the concentration of unknown samples by comparing the unknown to a series of samples of known concentrations. In the case of ethanol, a known peak is selected

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